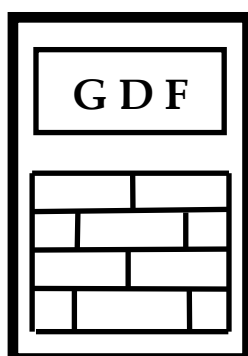


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Content

	no. pages
AC electric conductivity of untreated and mentally treated electrolyte aqueous solutions	6
About the author	1
Previous issues of GDF DATABANKS BULLETIN Errata	3

10+3 pages

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AC electric conductivity of untreated and mentally treated electrolyte aqueous solutions

In a previous study ac electric conductivity was performed on aqueous solutions of several ionic salts by considering topoenergetic working principles in view to evidence the effect of local tap water and the location of performing the measurements on structuring process [1]. On the other hand, recent series of studies on freezing process of water clearly revealed the mental field as its main driving potential [2]. In this way, study of ac electric conductivity on initial and mentally treated electrolyte aqueous solutions represents a natural result.

Experimental details: Saturated solution of crystalline $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ (1.080 g/mL) and nearly saturated anhydrous Na_2CO_3 (1.132 g/ mL) at 22 ± 2 °C were considered by using freshly boiled and quenched tap water as solvent. Procedure for ac electric conductivity was previously described [1]. Measurements at frequencies f : 100, 120, 1000 and 10,000 Hz were performed at the same temperature on successively diluted solutions with the coefficient 0.75 and saturated solutions are considered as having unity (1) concentration in arbitrary units (a.u.).

The mental treatment was applied by using the mental antenna recently described [2] and without special training and mental preparation before. The free end of it was immersed in the solution under test during the measurements. It is important to mention that all measurements were performed in the same place as the water freezing experiments [2], namely me as the operator being isolated in an area of approximately 200 m in diameter.

Finally, there were four series of measurements to be compared in view to reveal differences between them, namely as described in the following Table 1:

Table 1

symbols used for simplicity	solutions
anh	Na_2CO_3 anhydrous
CRST	$\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$
anh-MT	Na_2CO_3 anhydrous – Mentally Treated
CRST-MT	$\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ – Mentally Treated

Results: Figures 1-4 show some results for R and C as a function of [solute] which are similar with the previous results obtained for other electrolyte aqueous solutions [1]. Figures 5-14 shows the variations of phylogenic parameters (N, M, fo, -M/N, $-\text{N}^2/\text{M}$) from Universal representation of R and C as a function of frequency. It can observe at once that C is a more sensitive response eigenvalue at solute nature and MT than R. I have considered anhydrous and crystalline forms of Na_2CO_3 taking into account the differences of solutions structure. This can be observed also in the density values. I have chosen the bellow saturation solution of anhydrous species in view to avoid

crystallization. However, it has a bigger density than crystalline species due by a compact and crosslinked structure. I have expected that anh solutions have less mobility of dipoles than CRST solutions.

Figures 15 and 16 show the variation of $f \cdot R \cdot C = f/f_p$ on [solute] [1] at 100 Hz and 10 kHz. $f_p = 1/(R \cdot C)$ is the proper frequency of dipoles and the ratio f/f_p is always under unity because $f_p > f$. The above assumption can be substantiated by the following order of the solutions for this ratio:

$$\begin{aligned} 100 \text{ Hz :} & \quad \text{anh-MT} \geq \text{CRST-MT} > \text{CRST} > \text{anh} & (1) \\ 10 \text{ kHz:} & \quad \text{anh} > \text{anh-MT} \geq \text{CRST} > \text{CRST-MT}. \end{aligned}$$

Phylogenic parameters (M , $-M/N$, $-N^2/M$) have a sigmoidal variation on [solute] and they show distinct differences between the four solution series in the range of [solute]: 0.1-0.6 a.u., namely:

$$\begin{aligned} M \sim \text{LN}(\text{Ctr}): & \quad \text{CRST-MT} > \text{anh-MT} > \text{CRST} \geq \text{anh} & (2) \\ -M/N \sim \text{LN}(\text{ctr}): & \quad \text{anh} > \text{CRST} \geq \text{anh-MT} > \text{CRST-MT} \\ -N^2/M \sim \text{CS}: & \quad \text{anh} > \text{CRST} \geq \text{anh-MT} > \text{CRST-MT}. \end{aligned}$$

These relationships are also substantiated by the values in Table 2.

Table 2. Estimation of phylogenic parameters, y , for $x = [\text{solute}] = 0.3$.

$\text{LN}(C) = N \cdot \text{LN}(f - f_0) + M$ and $y = (a+b \cdot x)/(c+x)$ (correlation coefficients have values over 0.995 for all parameters y).

y	anh	CRST	anh - MT	CRST - MT
M	11.5 ± 0.2	12.3 ± 0.9	12.5 ± 0.3	13.2 ± 0.3
$-M/N$	23.80 ± 0.09	21.5 ± 0.1	21.46 ± 0.06	19.9 ± 0.1
$-N^2/M$	$-(0.021 \pm 0.002)$	$-(0.026 \pm 0.003)$	$-(0.028 \pm 0.003)$	$-(0.330 \pm 0.005)$

Conclusions:

1. Clear structural differences have been observed between untreated and mentally treated electrolyte aqueous solutions.
2. Mentally treated solutions have bigger dipole density with smaller volume/mass and coupling strength than untreated solutions.
3. Anh solutions have a crosslinked structure with smallest dipole density, biggest volume/mass and coupling strength of them with the inert lattice explaining fastest reaction at low frequency and lowest reaction at high frequency.

References

- [1] Gh. Dragan, Topoenergetic aspects of water structuring as revealed by ac electric conductivity, GDF Databanks Bull., 15(3), 2011.
[2] Gh. Dragan, DTA study of water freezing. V. Effect of a mental antenna, GDF Databanks Bull., 17(2), 2013.

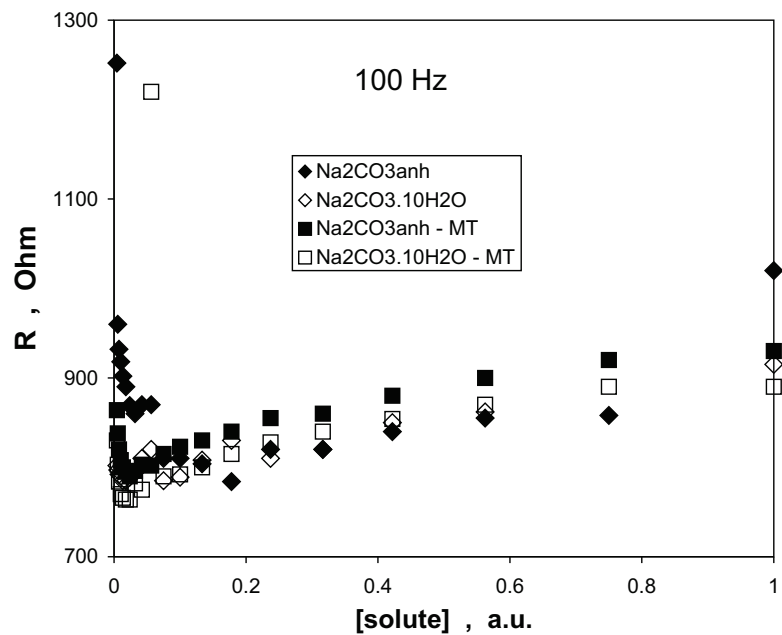


Figure 1.

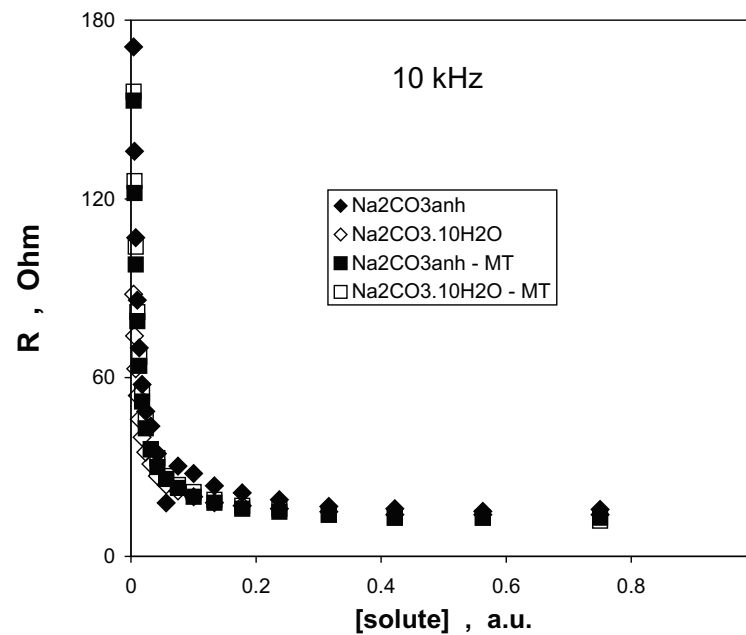


Figure 2.

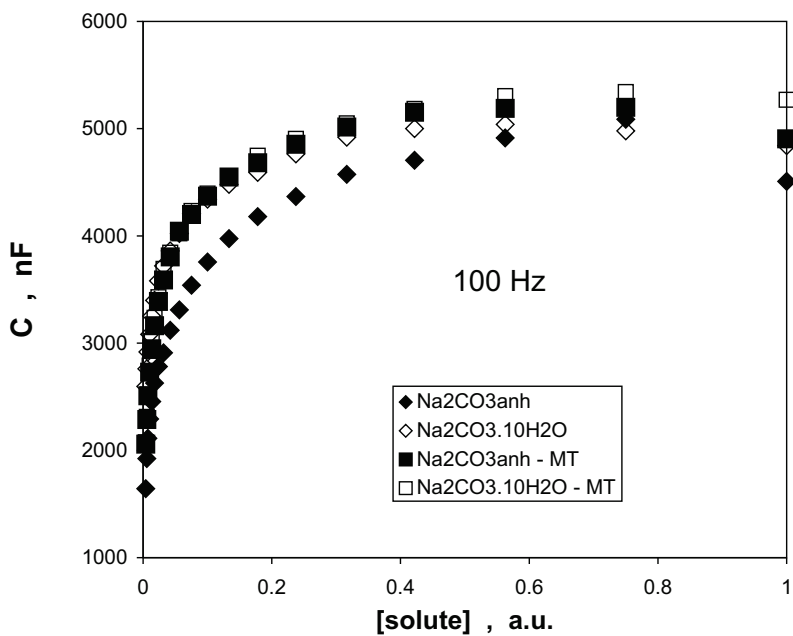


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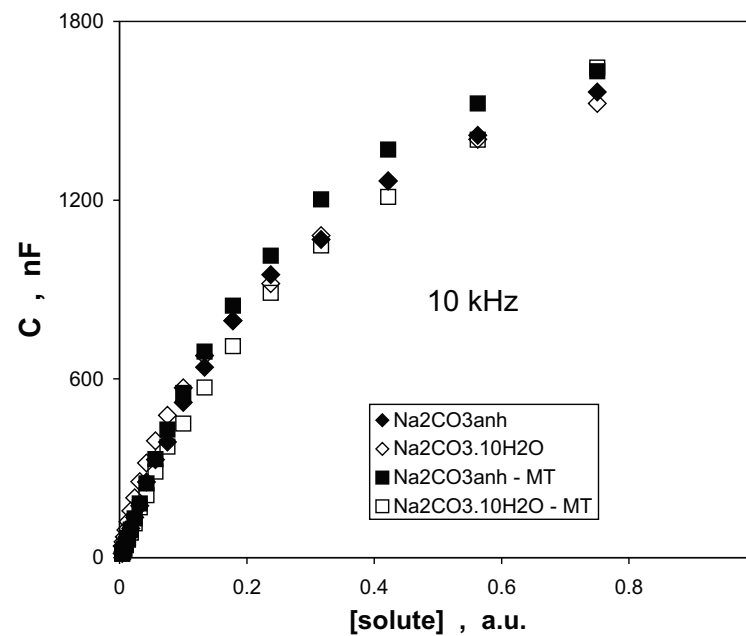


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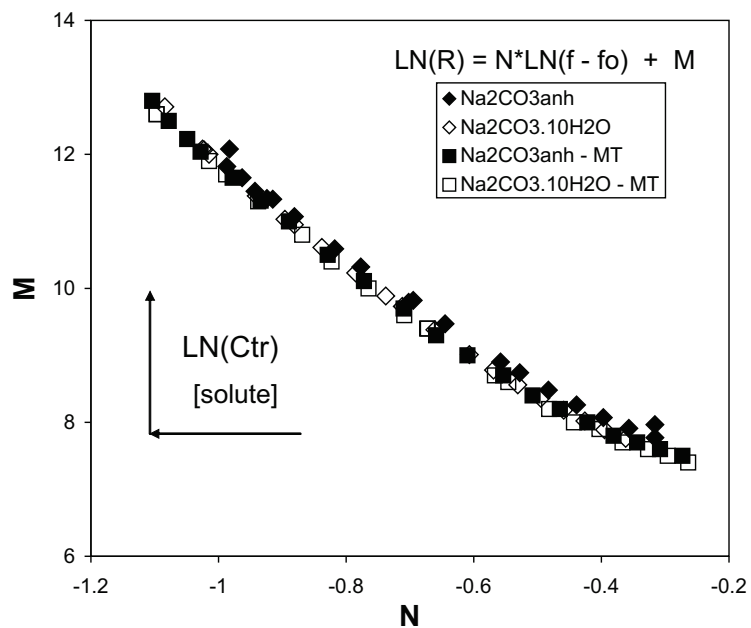


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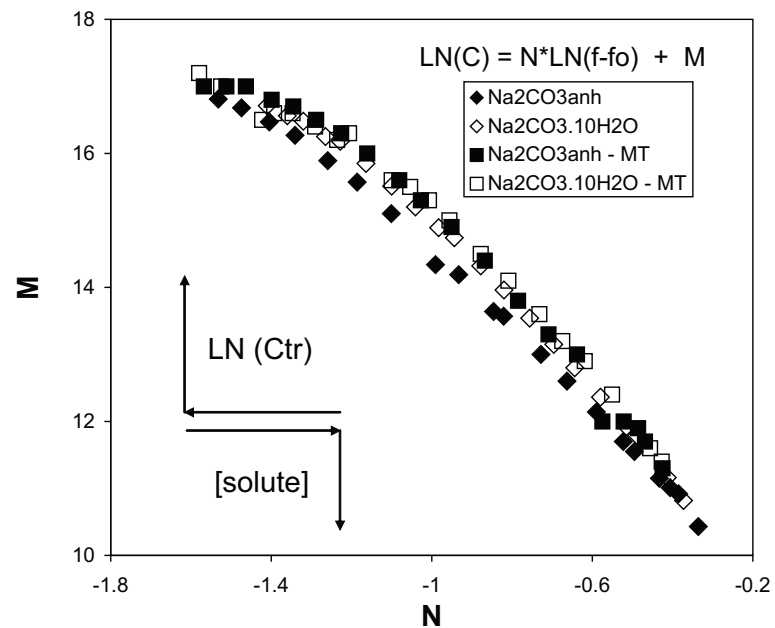


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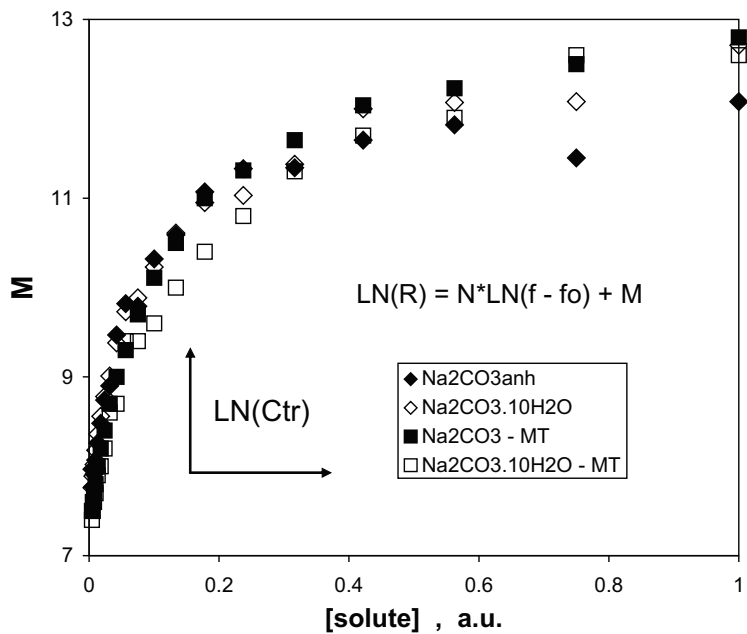


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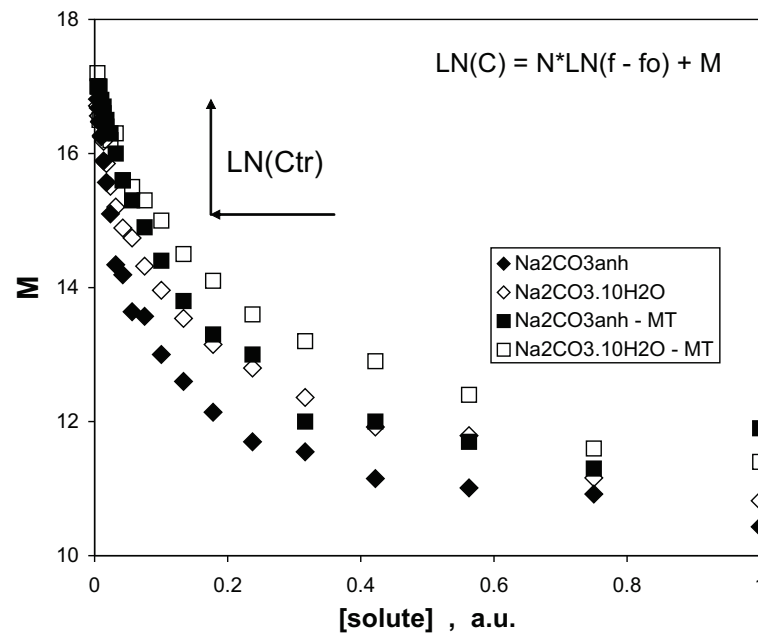


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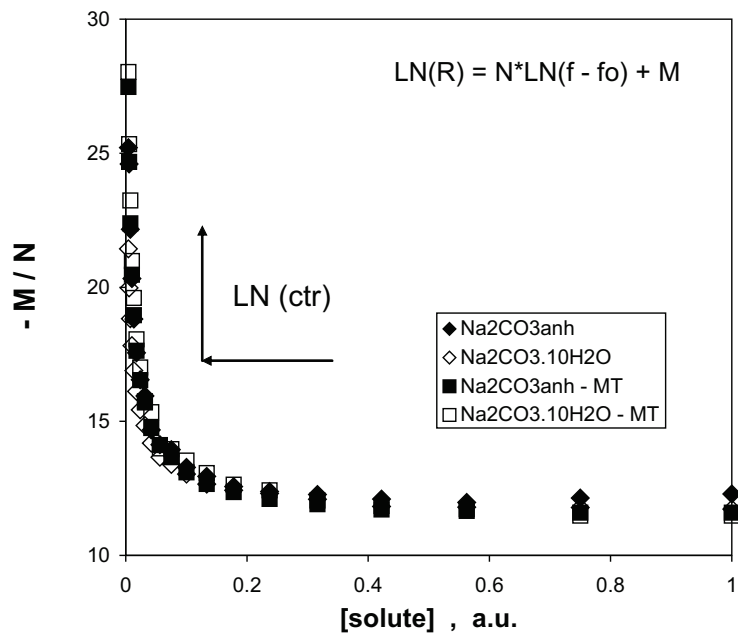


Figure 9.

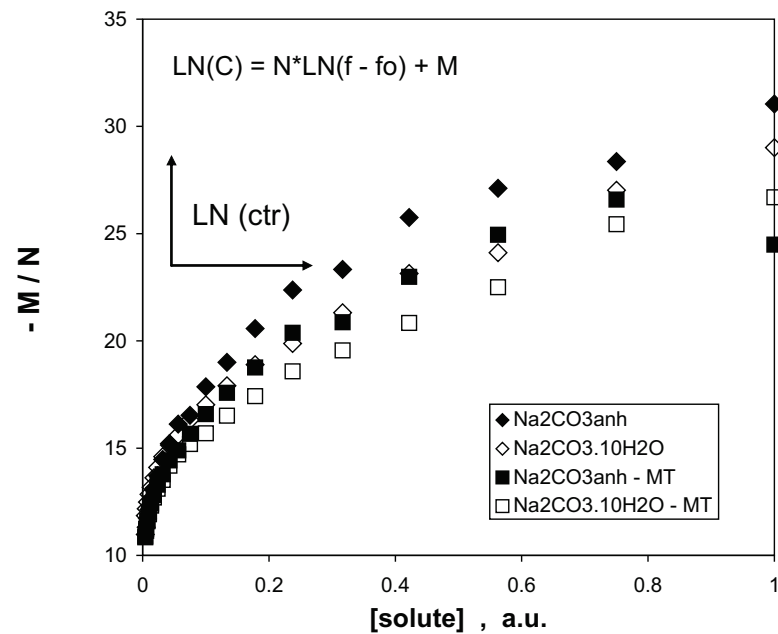


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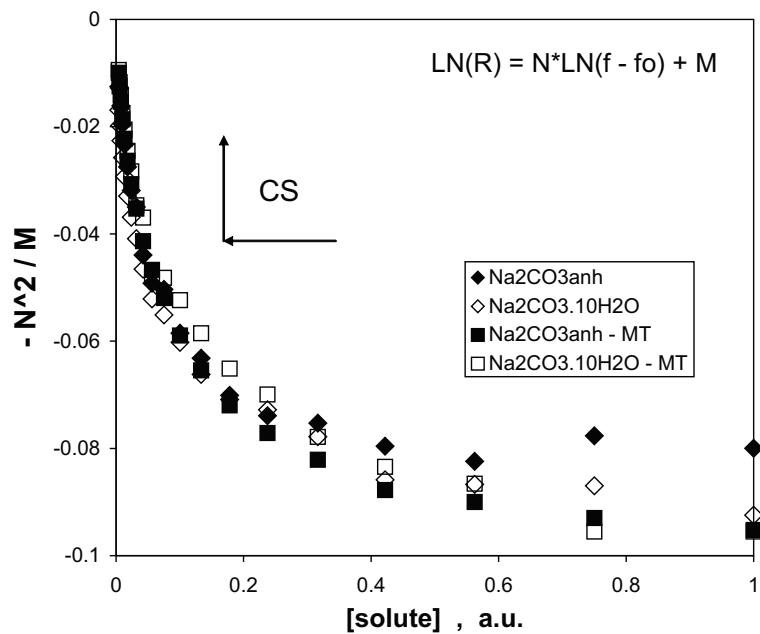


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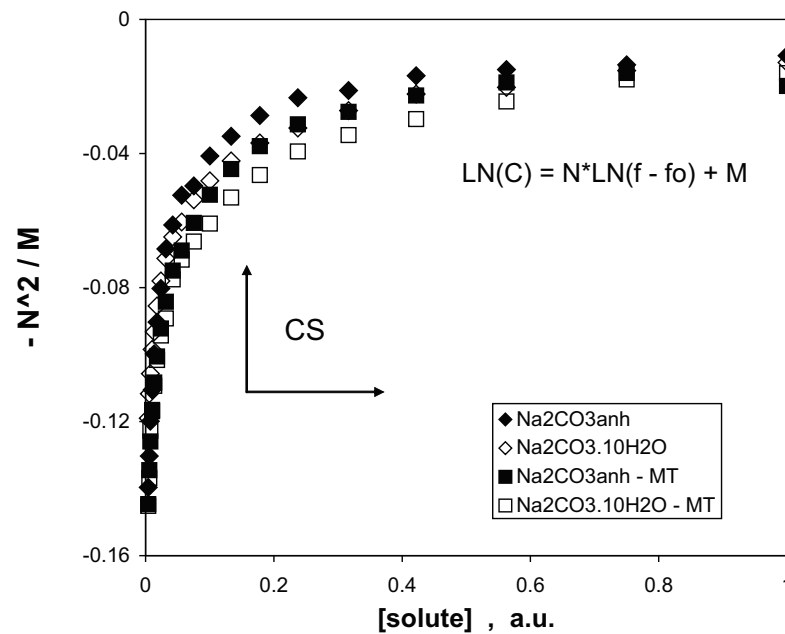


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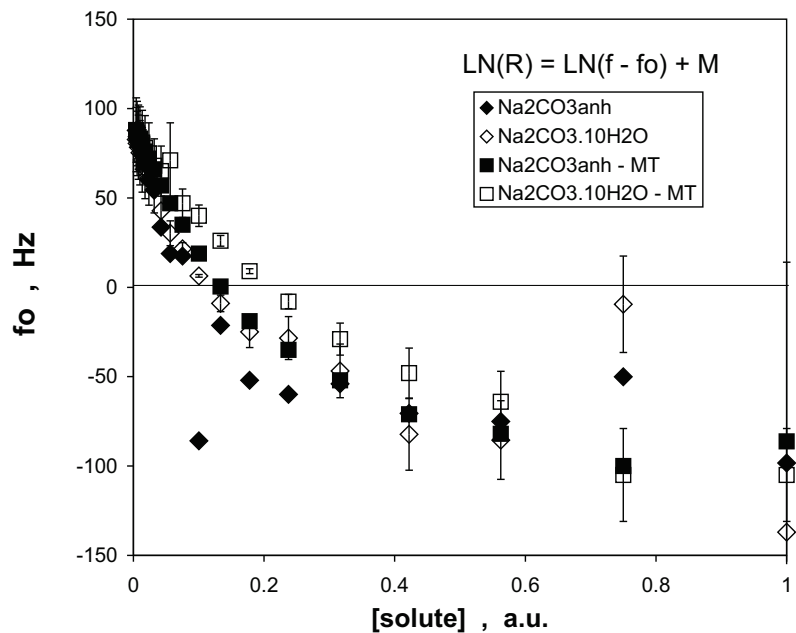


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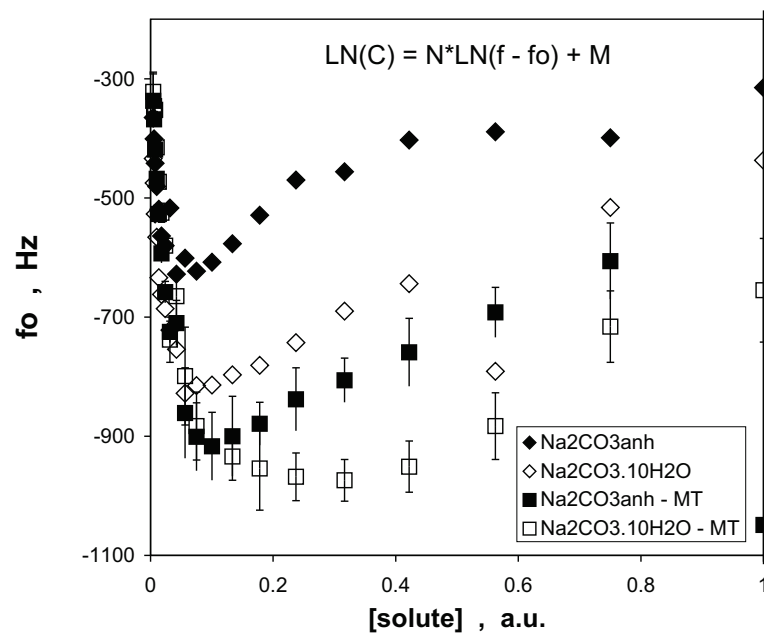


Figure 14.

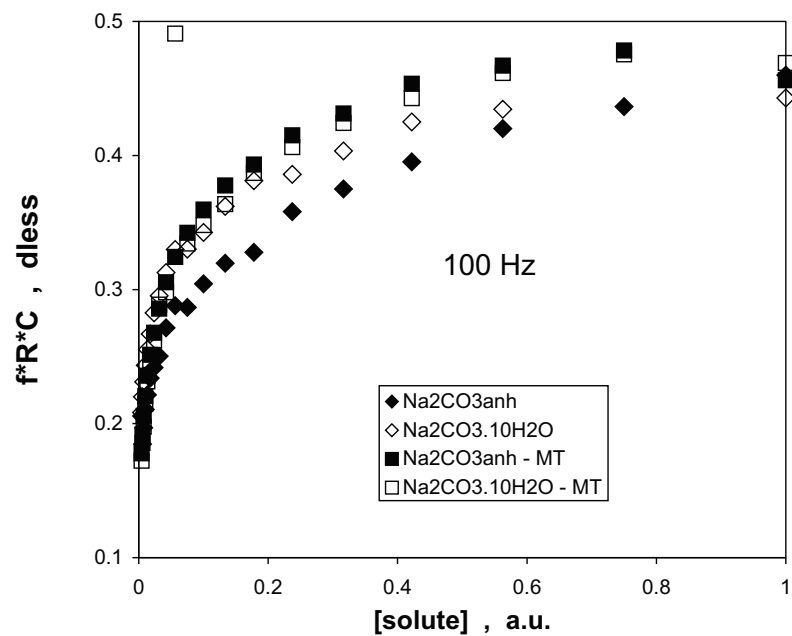


Figure 15.

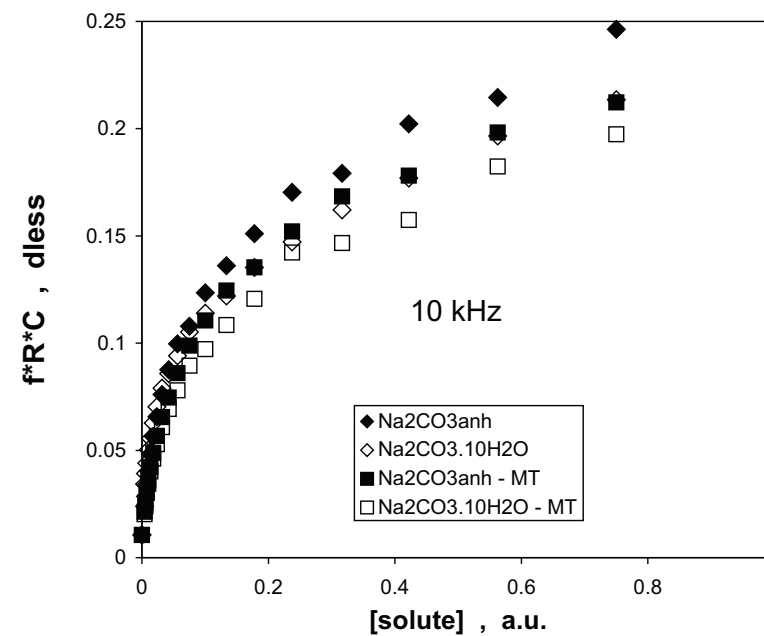


Figure 16.

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1997	1	2	Guide of good practice in metrology (Romanian)	AFI
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1998	2	2	Practical course of metrology (Romanian)	AFI
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2000	4	2	Measurement and Calibration.	AFI
2001	5	1	Editorial: Metrology ensures moral and technological progress. Topoenergetic aspects of amorphous-crystalline coupling. I. Composite behavior of water and aqueous solutions (paper presented at nanotubes and Nanostructures 2001, LNF, Frascati, Rome Italy, 17-27 October 2001). Events: Nanotubes and nanostructures 2000.School and workshop, 24 September – 4 October 2000, Cagliari, Italy.	F
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2004	8	2	Aspects of correct measurements of temperature. I. measurement of a fixed point according to ITS-90. Physics and Homoeopathy: some physical requirements for homoeopathic practice.(Plenary lecture at the 19 th SRH National Congress, 21-22 September 2004, Bucharest, Romania)	F
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2005	9	2	A new technique for temperature measurement and calibration. National Society of Measurements (NSM). Important warning for T-calibrator users: MSA has chose metrology well calibrators from Fluke (Hart Scientific).	F
2005	9	3	Universal representation of Cancer Diseases. 1. First sight on NSW-2003 report. Universal representation of Cancer Diseases. 2. UK cancer registrations on 1999-2002. Vital Potential can estimate our predisposition for cancer diseases.	F
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